

GCSE Science A / Physics

PH1HP Mark scheme

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Version/Stage: 1.0 Final

Mark schemes are prepared by the Lead Assessment Writer and considered, together with the relevant questions, by a panel of subject teachers. This mark scheme includes any amendments made at the standardisation events which all associates participate in and is the scheme which was used by them in this examination. The standardisation process ensures that the mark scheme covers the students' responses to questions and that every associate understands and applies it in the same correct way. As preparation for standardisation each associate analyses a number of students' scripts. Alternative answers not already covered by the mark scheme are discussed and legislated for. If, after the standardisation process, associates encounter unusual answers which have not been raised they are required to refer these to the Lead Assessment Writer.

It must be stressed that a mark scheme is a working document, in many cases further developed and expanded on the basis of students' reactions to a particular paper. Assumptions about future mark schemes on the basis of one year's document should be avoided; whilst the guiding principles of assessment remain constant, details will change, depending on the content of a particular examination paper.

Further copies of this mark scheme are available from aqa.org.uk

Mark Scheme

Information to Examiners

1. General

The mark scheme for each question shows:

- the marks available for each part of the question
- the total marks available for the question
- the typical answer or answers which are expected
- extra information to help the Examiner make his or her judgement and help to delineate
 what is acceptable or not worthy of credit or, in discursive answers, to give an overview
 of the area in which a mark or marks may be awarded
- the Assessment Objectives and specification content that each question is intended to cover.

The extra information is aligned to the appropriate answer in the left-hand part of the mark scheme and should only be applied to that item in the mark scheme.

At the beginning of a part of a question a reminder may be given, for example: where consequential marking needs to be considered in a calculation; or the answer may be on the diagram or at a different place on the script.

In general the right-hand side of the mark scheme is there to provide those extra details which confuse the main part of the mark scheme yet may be helpful in ensuring that marking is straightforward and consistent.

2. Emboldening

- 2.1 In a list of acceptable answers where more than one mark is available 'any **two** from' is used, with the number of marks emboldened. Each of the following bullet points is a potential mark.
- **2.2** A bold **and** is used to indicate that both parts of the answer are required to award the mark.
- **2.3** Alternative answers acceptable for a mark are indicated by the use of **or**. Different terms in the mark scheme are shown by a /; eg allow smooth / free movement.
- **2.4** Any wording that is underlined is essential for the marking point to be awarded.

3. Marking points

3.1 Marking of lists

This applies to questions requiring a set number of responses, but for which students have provided extra responses. The general principle to be followed in such a situation is that 'right + wrong = wrong'.

Each error / contradiction negates each correct response. So, if the number of errors / contradictions equals or exceeds the number of marks available for the question. no marks can be awarded.

However, responses considered to be neutral (indicated as * in example 1) are not penalised.

Example 1: What is the pH of an acidic solution? (1 mark)

Student	Response	Marks awarded
1	green, 5	0
2	red*, 5	1
3	red*, 8	0

Example 2: Name two planets in the solar system. (2 marks)

Student	Response	Marks awarded
1	Neptune, Mars, Moon	1
2	Neptune, Sun, Mars,	0
	Moon	

3.1 Use of chemical symbols / formulae

If a student writes a chemical symbol / formula instead of a required chemical name, full credit can be given if the symbol / formula is correct and if, in the context of the question, such action is appropriate.

3.2 Marking procedure for calculations

Full marks can be given for a correct numerical answer, without any working shown.

However, if the answer is incorrect, mark(s) can be gained by correct substitution / working and this is shown in the 'extra information' column or by each stage of a longer calculation.

3.3 Interpretation of 'it'

Answers using the word 'it' should be given credit only if it is clear that the 'it' refers to the correct subject.

3.4 Errors carried forward

Any error in the answers to a structured question should be penalised once only.

Papers should be constructed in such a way that the number of times errors can be carried forward is kept to a minimum. Allowances for errors carried forward are most likely to be restricted to calculation questions and should be shown by the abbreviation e.c.f. in the marking scheme.

3.5 Phonetic spelling

The phonetic spelling of correct scientific terminology should be credited **unless** there is a possible confusion with another technical term.

3.6 Brackets

(....) are used to indicate information which is not essential for the mark to be awarded but is included to help the examiner identify the sense of the answer required.

3.2 Ignore / Insufficient / Do not allow

Ignore of insufficient is used when the information given is irrelevant to the question or not enough to gain the marking point. Any further correct amplification could gain the marking point.

Do **not** allow means that this is a wrong answer which, even if the correct answer is given, will still mean that the mark is not awarded.

4. Quality of Communication and levels marking

In Question **2a** students are required to produce extended written material in English, and will be assessed on the quality of their communication as well as the standard of the scientific response.

Students will be required to:

- use good English
- organise information clearly
- use specialist vocabulary where appropriate.

The following general criteria should be used to assign marks to a level:

Level 1: basic

- Knowledge of basic information
- Simple understanding
- The answer is poorly organised, with almost no specialist terms and their use demonstrating a general lack of understanding of their meaning, little or no detail
- The spelling, punctuation and grammar are very weak.

Level 2: clear

- Knowledge of accurate information
- Clear understanding
- The answer has some structure and organisation, use of specialist terms has been attempted but not always accurately, some detail is given
- There is reasonable accuracy in spelling, punctuation and grammar, although there may still be some errors.

Level 3: detailed

- Knowledge of accurate information appropriately contextualised
- Detailed understanding, supported by relevant evidence and examples
- Answer is coherent and in an organised, logical sequence, containing a wide range of appropriate or relevant specialist terms used accurately.
- The answer shows almost faultless spelling, punctuation and grammar

Question	Answers	Extra information	Mark	AO / Spec. ref.
1(a)	chemical	correct order only	1	AO1
	kinetic		1	P1.2.1a
	sound		1	
1(b)	37.5% or 0.375	an answer rounded to 0.38 or 38% scores both marks an answer of 0.375 with a unit gains 1 mark an answer of 0.375% gains 1 mark an answer of 37.5 with or without a unit gains 1 mark allow 1 mark for an answer 3/8 3/8 with a unit scores 0	2	AO2 P1.2.1d
Total			5	

Question	Answers		Extra inf	formation	Mark	AO / Spec. Ref.	ID
2 (a)	scientific response. Examiners should also refer to the information on page 4 and apply a 'best-fit' approach to the				AO2 AO3 P1.2 HSW	E	
0 marks	Level 1 (1-2 marks)	Level	2 (3-4 marks)	Level 3 (5-6 r	narks)		
No relevant information	There is a relevant statement about an energy saving method	clear c energy method cost ef	is at least one omparison of saving ds and their fectiveness appropriate atton	There is a comp energy saving m and their cost effectiveness wit appropriate calc Comparison to in further detail.	ethods th ulations.		
examples o response	f physics points made i	n the	extra informati	on			
energy ef (energy / loft insular double-gl examples or effectivenes loft insular the long t double-gl loft insular time double-gl time payback to examples or for cost er following double-gl don't instarpurposes double-gl lifetime	tion is the most cost effective erm azing is the least cost effection has the shortest pay azing has the longest pay time calculated for any metric further detail are fectiveness install in the order: loft, cavity wall, bo azing a energy use install in the order: boiler, loft, cavity wazing all double-glazing for insurance.	tall istall le cost ctive in ective back /back ethod iller, ne /all, ulation	loft insulation: 2 double glazing:				

Question 2 continued . . .

Question	Answers	Extra information	Mark	AO / Spec. ref.
2(b)(i)	how effective a material is as an insulator	accept 'heat' for energy accept how effective a material is at keeping energy in accept the lower the U-value the better the insulator accept the lower the U-value the lower the rate of energy transfer	1	AO1 P1.1.4a
2(b)(ii)	(the U-value) decreases		1	AO1 P1.1.4b
Total			8	

Question	Answers	Extra information	Mark	AO / Spec. ref.
3(a)(i)	0.5 °C		1	AO3 P1.1
3(a)(ii)	data is continuous or both variables are continuous or independent variable is continuous or time is continuous	accept results / measurements for data accept data is not categoric one variable is continuous is insufficient air temperature is continuous is insufficient	1	AO3 P1.1
3(b)(i)	20.5 (°C)		1	AO2 P1.1
3(b)(ii)	60 (minutes)	accept 1 hour	1	AO2 P1.1
3(c)(i)	so a comparison can be made or outside temperature is a control variable	accept: (outside) temperature would affect energy required (to maintain temperature of the house) or (outside) temperature would affect internal temperature (of the house) or heat loss will be faster on a cold day outside temperature will affect the results is insufficient fair test is insufficient	1	AO3 P1.1

Question 3 continues on the next page . . .

Question	Answers	Extra information	Mark	AO / Spec. ref.
3(c)(ii)	the cost is equal to the number of kWh x the cost per kWh	accept (heating) bill depends on (number of) kWh used accept energy for kWh	1	AO3 AO2 P1.1
	calculation 0.8 / 8.0 = 0.1 or 10%	allow 7.2 / 8.0 = 0.9 or 90%	1	
3(c)(iii)	heating is on for more / less time (than anticipated)		1	AO3 P1.1
	because some days it is cooler / warmer (than anticipated)	accept other sensible suggestions	1	
		 an answer giving two sensible situations gains 2 marks possible examples: some houses have different amounts of insulation there are different styles of house 		
		temperature (inside / outside) is always changing is insufficient		
Total			9]

Question	Answers	Extra info	rmation	Mark	AO / Spec. ref.
4(a)(i)	microwave			1	AO1 P1.5.1k
4(a)(ii)	refraction			1	AO1 P1.5.1g
4(b)(i)	wave M continues as a straight line ionosphere and shown reflected correctly reflected wave shown as a reaching the top of the receiver		accept reflection at or within the ionosphere if more than 2 rays shown 1 mark maximum ignore arrows	1	AO2 P1.5.2b
4(b)(ii)	normal drawn at point where their ionosphere Saturation Transmitter The Earth	I meets the tellite		1	AO1 P1.5.2a

Question 4 continues on the next page. . .

Question 4 continued...

Question	Answers	Extra information	Mark	AO / Spec. ref.
4(c)	any two from: • transverse • same speed (through air)	accept speed of light or 3 x 10 ⁸ m/s	2	AO1 P1.5.1a,bc, d,g
	 can be reflected can be refracted can be diffracted can be absorbed transfer energy can travel through a vacuum can be polarised show interference 	an answer travel at the same speed though a vacuum scores 2 marks		
		travel in straight lines is insufficient		
Total			7	

Question	Answers	Extra information	Mark	AO / Spec. ref.
5(a)(i)	infrared (radiation)	accept IR (radiation)	1	AO1 P1.1.1b
5(a)(ii)	(heated) water turns to steam steam turns a turbine turbine turns a generator	ignore reference to fossil fuels do not accept water evaporates to steam accept turbine connected to a generator	1 1 1	AO1 P1.4.1a
5(b)(i)	(so the molten salts) can store large amounts of energy	accept there is a small temperature change for a large energy transfer accept heat for energy	1	AO2 P1.1.4d
5(b)(ii)	16 (hours)	an answer that rounds to 16 gains 2 marks eg 15.71 allow 1 mark for a correct substitution ie 2 200 000 = 140 000 x t	3	AO2 AO1 P1.3.1c
5(b)(iii)	the number of daylight hours varies the (mean) power (received from the Sun per square metre) varies	less sunlight is insufficient accept an answer in terms of maximum possible electrical output only possible during Summer for 1 mark	1	AO2 AO3 P1.4

Question 5 continues on the next page . . .

Question 5 continued . . .

Question	Answers	Extra information	Mark	AO / Spec. ref.
5(c)(i)	non-renewable power stations have higher Capacity Factors than renewable power stations		1	AO1 AO3 P1.4
	fuel (for non-renewable power stations) is always available	reference to non-renewable power stations operating all the time is insufficient non-renewable energy sources are reliable is insufficient	1	
	(most) renewable energy sources are unpredictable / unreliable	accept (most) renewable energy sources depend on the weather	1	
5(c)(ii)	the (proportion of) time that solar storage power stations can generate electricity is greater (than for other renewable energy sources)		1	AO1 P1.4
Total			14	

Question	Answers	Extra information	Mark	AO/Spec ref.
6(a)	Doppler (effect)		1	AO1 P1.5.4a
6(b)	(reflected microwaves) wavelength decreased (reflected microwaves) frequency increased (reflected microwaves) have same speed		1 1 1	AO1 P1.5.4a P1.5.1d
Total			4	

Question	Answers	Extra information	Mark	AO / Spec. ref.
7(a)	4200	allow 2 marks for correct substitution ie 6930 = 0.330 x c x 5.0	3	AO2 AO1 P1.1.4d
		answers of 1050 or 840 or correctly calculated answer from correct substitution of incorrect temperature change or		
		identification of temperature change ie 5 °C gain 1 mark		
	J/kg°C	accept J/kg K	1	
7(b)	(in a metal) free electrons	to gain full credit the answer must be in terms of free electrons	1	AO1 P1.1.3a
	gain kinetic energy	accept move faster	1	
	(free electrons) transfer energy to other electrons / ions / atoms	do not accept particles	1	
	by collision		1	
		allow a maximum of 2 marks for answers in terms of atoms / ions / particles • gaining kinetic energy or vibrating faster / more		
		transferring energy by collisions		
7(c)	(air) particles spread out		1	AO1 P1.1.3a
	(which causes the) air to become less dense / expand	do not accept particles become less dense	1	
	(so the) warm air rises	do not accept heat rises particles rise is insufficient	1	
7(d)	large surface area	ignore references to type of	1	AO1
	black / dark (colour)	metal or external conditions	1	P1.1.3c
Total			13	